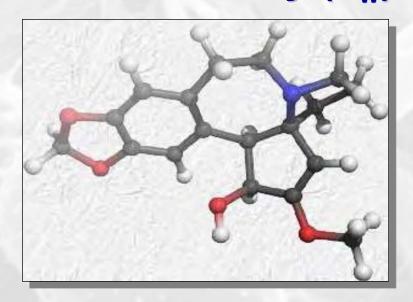






# Theory of Water Activity (a<sub>w</sub>)



**Quality control of food!!** 







## **AGENDA**

- 1. Definition of water activity
- 2. Hurdle technology
- 3. Microbial growth
- 4. Measurement of water activity
- 5. Sorption isotherms
- 6. Novasina product range











Raoult's law:

$$p = \gamma \cdot p_S$$

p: Partial vapour pressure

y: mole fraction

p<sub>S</sub>: Saturation pressure



chemical potential of water in a real solution:

$$\mu_W = \mu_W + R \cdot T \cdot \ln a_W$$

thus the Raoul's law becomes:  $p = p_S \cdot a_W$ 

the osmotic pressure is given as:

$$V_W \cdot \pi = R \cdot T \cdot \ln a_W$$

V<sub>w</sub>: molar volume of water

 $\pi$ : osmotic pressure

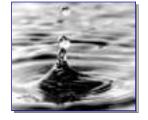
The activity  $a_W$  of an aqueous solution is called **water activity**.





### **Definition of water activity**

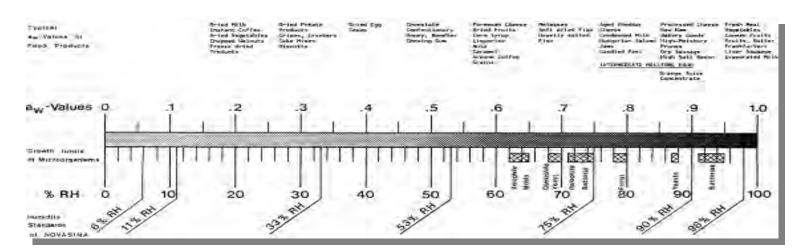
The water activtiy is defined as:



**a**<sub>w</sub> = water vapour pressure over sample\* saturation vapour pressure of pure water\*

\* at the same temperature °C

 $a_w = p/p_0 = equilibrium humidity = EHR (%) / 100$ 



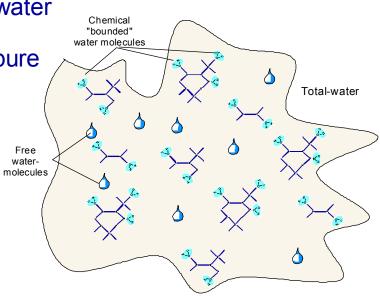




## a<sub>w</sub> = water activity



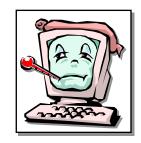
- The degree of availability of water is measured by water activity (a<sub>w</sub>)
- Water activity is a measurement of the energy status of the water in a system
- The water activity states how much the water vapour pressure of a product is reduced compared to the saturation pressure of pure water at the same temperature
- Water activity is the amount of "free" or "available" water in a product
- "Free" is also defined as:
   same physical properties as pure water







## Water activity (a<sub>w</sub>) and its influence



## Temperature → a<sub>w</sub>

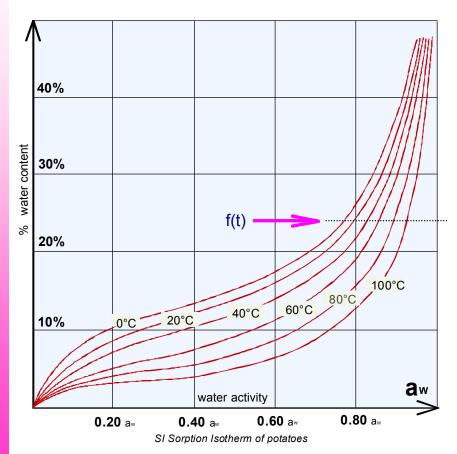
The water activity depends on temperature.

- The water activity of a product changes with its temperature!
- -> Thus changes in temperature can cause a water migration between different components





## **Temperature effect**



## Water activity changes with temperature thus:

- It is necessary to control temperature.
- Compensate for temperature difference between sample and sensor.

### **Purposes for temperature control:**

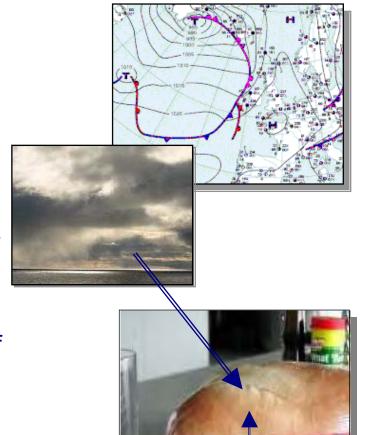
- Aw-measurement at a defined temperature
- Lab to lab sample comparison
- Isotherm determination
- Compliance with government or internal regulations
- Eliminate extreme ambient temperature fluctuations





## **Water migration**

- Two ingredients may have the same moisture content, but totally different a<sub>w</sub>-values.
- The water migrates from regions of high a<sub>w</sub> to regions of low a<sub>w</sub>, and not between areas of unequal moisture content!
- Water migration between different layers of a multicomponent food causes undesirable textural changes.
- This can be reduced by knowing and influencing the various levels of a<sub>w</sub>.

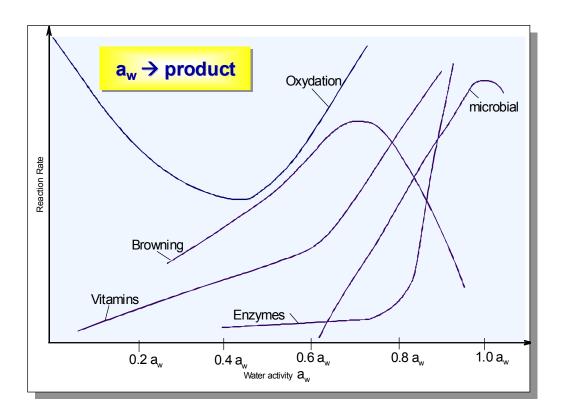






### Influence of the water activity on the product

Water activity is jointly responsible for the growth of undesirable organisms such as bacteria or fungi, which produce "toxins" or other harmful substances. But also chemical/biochemical reactions (e.g. Maillard reaction) increasingly take place



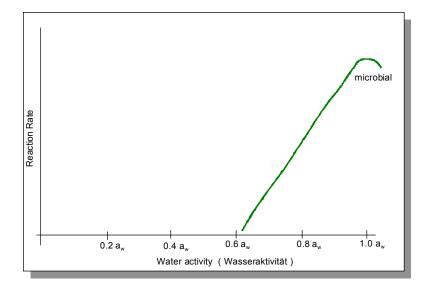
## The following properties of a product are affected:

- Mircrobiological stability
- Chemical stability
- Content of proteins and vitamins
- Colour, taste and nutritional value
- Stability and durability
- Solubility and texure





### **Microbial Growth**



- Decreasing quickly with declining aw-value.
- There is no more microbial growth under 0.6a<sub>w</sub>!





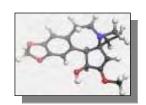






### **Enzyme Activity**

- For various enzymes a minimal water content is necessary
- It leads to changes in nutritional value, colour and flavor
- The most enzyme activity slows down below an a<sub>w</sub>-value of 0.8

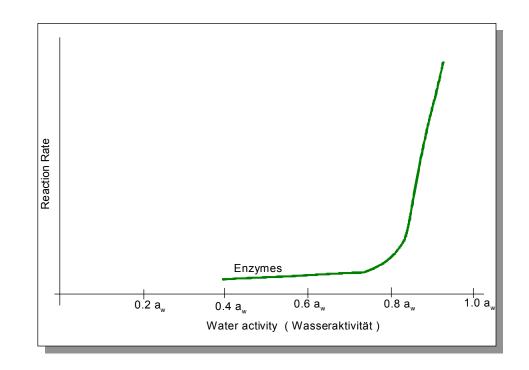


### **Enzyme activity** water effects:

- Dissolve substrate
- Increase substrate mobility
- Water can be a reactant

### **Enzyme stability** water influences:

- The denaturations:
  - Hydrolysis
  - Deamidation
  - Oxidation

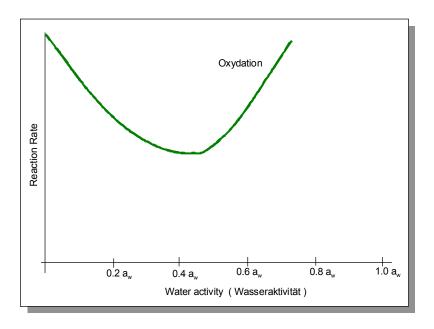






### **Lipid Oxidation**

- Oxidation is one of the major causes of food spoilage and it is initiated by metal ions.
- The reaction rate falls with a lower a<sub>w</sub>-value and increases again.
   The minimum is at 0.3 a<sub>w</sub>



## Causes of anti-oxitative effect (range $0 - 0.3 a_w$ )

- Reduced oxygen diffusion
- Less available metal ions due to bonding with water molecules
- Free radicals are bounded

## Causes of pro-oxidative effect (range 0.3 – 1 a<sub>w</sub>)

- Increased mobility of the reactants (metal ions and oxygen)
- Increased dissolution of catalysts
- The food swells what causes a surface extension



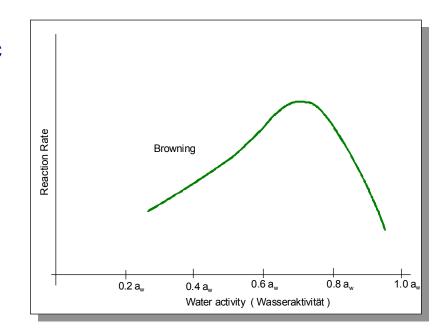


### **Non-enzymatic reactions**

For non-enzymatic reactions you can mainly mention the protein denaturation and the non enzymatic browning. Mostly the non-enzymatic browning causes the most noticeable changes. The extent of browning reactions depends on the water content repectively the **water activity** of a product.

The probability of the non-enzymatic reaction is more probable with increased  $a_w$ -value and reaches a :

Maximum at  $0.6 - 0.7a_w$ 

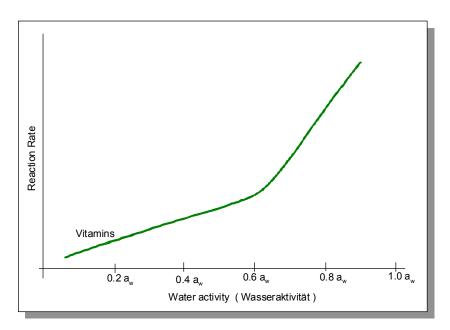






### **Nutrient Degradation**

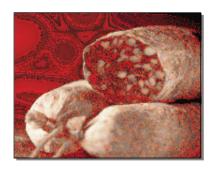
Vitamins are **essential food components** therefore its preservation during the processing of food is of prime importance.



The reaction rate of vitamins increases with **increasing aw-value**. This leads to degradation and loss of vitamins.









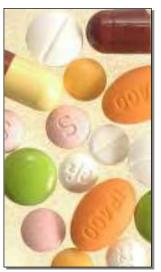




# Why is the measurement of the water activity (aw-value) in foods and pharmaceuticals so important?











### The importance of "a<sub>w</sub>" measurement :

	microbial safety	to predict shelf life
(m)	HACCP requirements	as part of QC
	fulfil government regulations	EU, USA, Japan
	control chemical reactions, physical properties	to guarantee product quality, good taste, colour, crispness













## **AGENDA**

- 1. Definition of water activity
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- 3. Microbial growth
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- 5. Sorption isotherms
- 6. Novasina product range







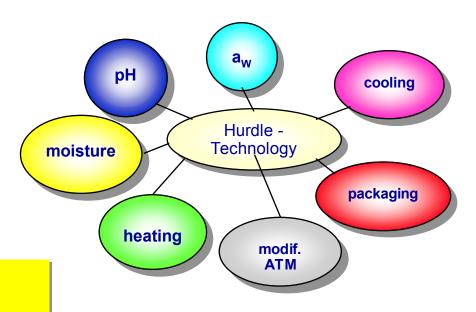
## "Hurdle-Technology" in food

Since centuries food was preserved by different methods. Often these methods are applied in combination.

The principles which form the basis of the traditional methods are made quantitative ascertainable by:

- f-value (heating)
- t-value (cooling)
- pH-value (acidification)
- aw-value (drying, salting, sugaring)
- eh-value (deoxygenation)
- compeditive flora (fermentation)
- preservatives

An intelligent application of a combination of these factors is called **Hurdle-Technology** 







## "Hurdle Technology"

(Illustration of the hurdle technology by 7 examples)

The symbols have the following meaning:

f : heating

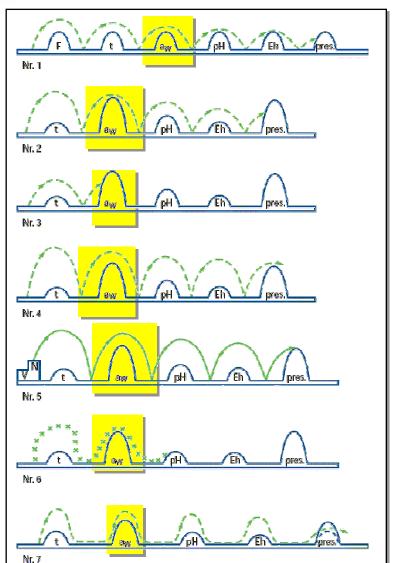
t : cooling

**a**<sub>w</sub>: water activity

pH: acidity

**eh** : redox potential

pres.: preservatives







## Measures to Lowering the water activity

- Sun-drying, adding salt or sugar are practices known for centuries
- Dehydration/freezing or drying are modern techniques
- Humectants; deacreasing availability by binding water; like honey or corn syrups (sugar), polyols (eg. glycerine, glycerol, propylene glycol, sorbitol), but also proteins, amino acids etc. are today commonly found in foods!
- A smart combination results in an optimum texture, flavour, taste, colour of the final product.











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### Requirements for the growth of microorganisms

### **Nutrients**

- Yeasts prefer food rich on carbonhydrates (dessert, sauces, fruit juces)
- Salmonella splits protein and grows on food (meat, sausage, fish, egg and cheese)

### **Temperature**

- Most microorganisms show an optimal growth between 15°C and 40°C
- The growth of mircroorganisms depends on temperature

### Water

- Not only the water content is responsible for the growth of microorganisms, but the free available fraction of the water (a<sub>w</sub>-value) is determing
- An high sugar- and salt content is responsable for an well binding of the free water into an probe. This is a possible way to bring the free water content down.







### Requirements for the growth of microorganisms

### pH value

- Bacterias prefer a lightly acid respectively neutral environment,
   pH-Wert 4.5 7
- For yeasts and molds a pH 4.5 is optimal

### Oxygen

- Molds and yeast need oxygen
- Certain bacterias (e.g. clostridium) grow without oxygen
- Other bacterias can adapt to changing environmental conditions





## Way of infection and sources of food poisoning

### **Primary contamination:**

Production animal is contaminated with pathogens (often without symptom of disease respectively pool of pathogens Salmonella, Campylobacter, EHEC, Shigella, ...)



### Secundary contamination:

Contamination during the winning of the food by smear infection via unclean equipment, lack of hygiene or insufficient care, ...
 (all pathogens possible: Intestinal bacterias, Clostridium, Staphylococcus)



- Food contamined by pathogens
- Food poisoning (by toxins)





## For innumerous food spoilage organisms a **minimum a<sub>w</sub>-value** is known!

Below this value its growth and toxin production is inhibited

Bacteria, for example:	Minimum aw-value	
Staphylococcus aureus :	0.86a <sub>w</sub>	
Clostridium botulinum A and Escherichia coli	0.95a <sub>w</sub>	
Salmonella	0.92a <sub>w</sub>	
Molds, for example.:		
Aspergillus flavus: produces toxin above 0.83a <sub>w</sub> , but does not grow below	0.78a <sub>w</sub> .	
Fungi, other molds, mildew, yeasts go lower, but <i>not beyond</i>	0.60a <sub>w</sub> .	





### **Growth** and sources of different bacterias

	meat	milk	poultry	egg	fish	fruits & vegetables
Salmonella	•	•	•	•		•
Pathogenic E. coli	•	•				•
Campylobacter		•	•			
Y. enterolytica	•		•			
C. perfringens	•		•			
C. botulinum	•				•	•
L. monocytogenes	•	•	•		•	•
Vibrio sp.						•
S. aureus	•	•	•			





## **a<sub>w</sub>-range** of **food** and its microflora

a <sub>w</sub> -range	food	microorganisms		
> 0.98	Fresh meat Fresh fish Fresh fruits & vegetables	Pseudomonas, Escherichia, Proteus, Shigella, Bacillus, Clostridium perfringens		
0.92 – 0.98	Sausages Cheese Bread	Salmonella, C. botulinum, Lactobacillus, Pediococcus, some yeasts and molds		





## **a<sub>w</sub>-range** of **food** and its microflora

a <sub>w</sub> -range	food	microorganisms	
0.92 – 0.87	Fermented sausages, Biscuits, cheese, margarine	Various yeasts (Cand Micrococcus	ida,)
0.87 – 0.80	Fruit juice concentrate, Condensed milk, chocolate syrup, flour, fruit cake, ham	Most molds, Staphylococcus aureumost Saccharomyces Debaryomyces	





## **a<sub>w</sub>-range** of **food** and its microflora

a <sub>w</sub> -range	food	microorganisms		
0.80 - 0.60	Dried fruit Spicies Cereals Nuts	Xerophilic fungi (Aspergillus candidus,) Osmophilic yeasts		
> 0.60	Confectionery Honey Noodles egg- and milk powder cookies, cracker, etc.	No microbial growth but may remain viable		





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## Measurement of water activity

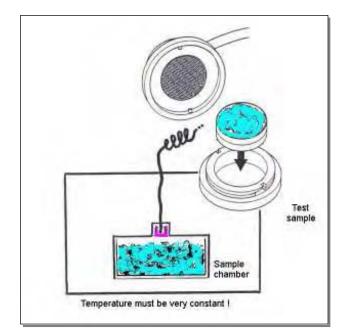
The "offline" measurement :

The test sample is placed in a completely sealed measuring chamber and the sample humidifies or dehumidifies the air volume inside the chamber till the **equilibrium humidity** is reached.

This exchange takes place due to the **partial** water vapour pressure difference between the sample and the air.

The measuring speed is largely dependent on the sample properties.



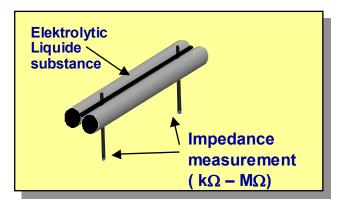






The humidity/aw measurement element

## **Resistive Electrolyte Cell**



- directly measuring the a<sub>w</sub>-value
- virtually hysteresis-free
- accurate to 0.003a<sub>w</sub> (0.3%rh), from below
   0.03a<sub>w</sub> up to 1.00a<sub>w</sub>
- excellent repeatability of 0.002a<sub>w</sub> (0.2%rh)
- very easy and simple to change a calibrated measurement element (full accuracy)
- simple to calibrate the aw-system with saturated salt solutions

#### **Function:**

Liquide electrolyte changes resistance when the humidity around is changed

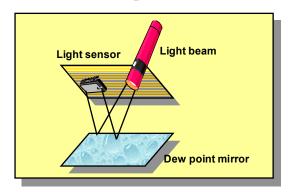








### The dew point measurement methode





Function:
Optical identification of condensation on mirror temperature

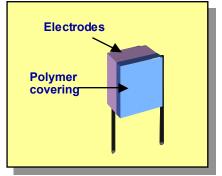
- needs a lot of electronics and control tools
- is not easy to handle and maintain (daily mirror cleaning)
- indirect measurement; gets a result by calculating the a<sub>w</sub> from both,
   mirror dew point and sample temperature
- tremendous errors can be caused by volatiles (eg. alcohols!) as well as from sample surface colour and structure, which influence the infrared surface temperature measurement
- results are sometimes calculated too fast, so repeated measurements are recommended and frequent mirror cleaning is advisable





### Das Feuchte Messelement die

**Capacitive measurement** 





#### **Function:**

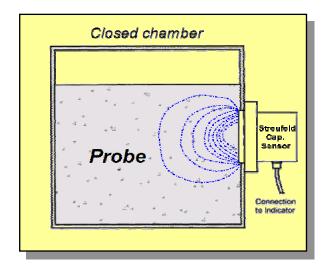
Polymer expands with increasing humidity, thus changing capacitance

- easy and simple to handle
- quick reading of (+/- 0.04-0.05 a<sub>w</sub>)
- directly measuring rh, have a physically given hysteresis of min. 1.5%rh (0.015a<sub>w</sub>)
- "second sorption effect" in the high range: they absorb additional water molecules, which leads to a higher reading
- Problematic against chemical contamination





## Capacitive stray field measurement (aw-cryometer)



- Simple to handle
- Max. immersion depth of few cm!
- Accurate and not affected by volatiles
- limited aw-range (>0.80....1.00 a<sub>w</sub>)
- Robust housing
- adjustable



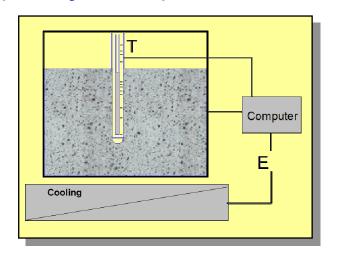
#### **Function:**

Changing of the capacitance of a stray field sensor by water in a sample





## Freezing point determination (aw-cryometer)





#### **Function:**

Accurate determination of the depression in freezing point relative to pure water

- simple to handle
- accurate and not affected by volatiles
- limited aw-range (>0.90aw)
- small sample size
- aw-value determined at the freezing point, the result is than extrapolated on 25°C.





### Saturated salt solutions

The saturated salt solutions can be used over many years as often as desired to verify and calibrate water activity instruments.



Advantages of saturated humidity standards:



- Simple handling
- Long term stability of the standards
- Simple control of lifetime







## Sample preparation

A special sample preparation is **not needed**, but larger samples should be cut into small pieces: a volume of 8...15ml (ccm) is sufficient. Depending on the product, crushing or grinding may change the  $a_w$ -value, so manually cutting and immediately testing is the safest procedure.

Multicomponent products may take a **very long time** to establish a final, common  $a_w$ -value (several days, even weeks!), so it might be advisable to separate the different components and measure them individually.









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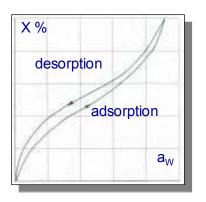
**Growing of** 

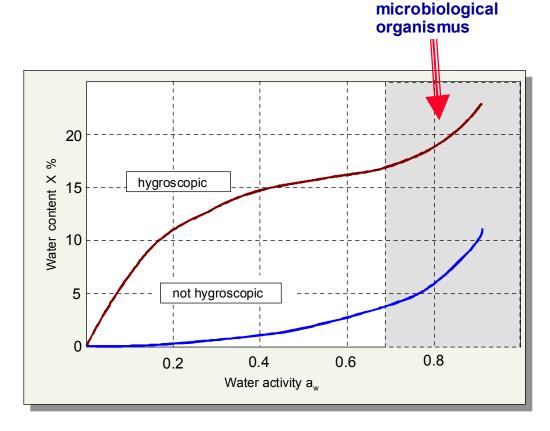




### **Record of the sorption isotherm**

- Relation between water content and water activity of a product
- Product specific and temperature dependent
- Desorption or adsorption isotherm (hysteresis)









### **Record of the sorption isotherm**

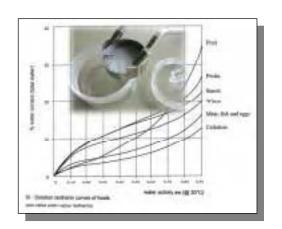
It is possible to record a sorption isotherm (SI curve) with the new water activity instrument **LabMaster-aw** thanks to:

- temperature controlled measuring chamber+/-0.2K
- temperature range 0°C ... 50°C
- 7 Novasina humidity standards













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## LabMaster-aw









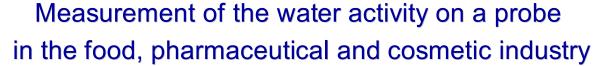
- Best accuracy (+/- 0.003 a<sub>w</sub>)
- High precision temperature controlled chamber
- Wide measurement range 0.03a<sub>w</sub> up to 1.00 a<sub>w</sub>
- Single or multi chamber version (1 LabMaster and max. 9 LabPartner)
- Precondition chamber for sample
- Large, backlighted LC display
- Simple to maintain and service
- 6 to 7 point calibration available (with Novasina SC standards)
- High temperature range (0....50°C and accuracy : +/-0.2°C)
- SI set to measure the sorption isotherm available















### **Products:**















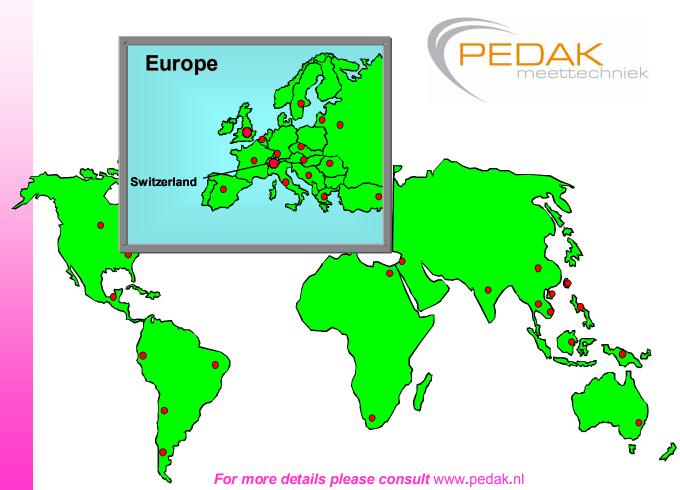




Israel



### International representatives :



#### worldwide

**Argentina Australia Austria** Belgium **Brazil** Canada Chile China Columbia Czech Rep. Denmark **Egypt England Finland** France Germany Greece Holland **Hong Kong** Hungary India Indonesia

Iran

Italy Japan Korea Malaysia Mexico **New Zealand Norway Philippines Poland Singapore** Slovakia South Africa **Spain** Sweden **Switzerland** Taiwan **Thailand Turkey USA** 

Vietnam





### International references:

